

ENGY 3310 Fundamentals of Nuclear Science and Engineering
Spring 2016

HW #5: Various Mass-Energy-Momentum Relationships

Problem 1 Mass-Energy Equivalency (10 points)

An electron starting from rest is accelerated across a potential difference of 5 million volts.

- a. What is its final kinetic energy (MeV)?
- b. What is its total energy?
- c. What is its final mass?

Problem 2 Speed and Momentum of Various Particles (10 points)

Compute the speed (v/c) and momentum (N-s) of the following particles:

- a. 1 MeV neutron
- b. 1 MeV electron
- c. 1 MeV gamma ray

Problem 3 Wave-Particle Duality (5 points)

We often refer to gamma radiation as a "gamma ray particle" (as in the previous problem).

Compute the de Broglie wavelength for a 1 MeV gamma ray and discuss whether it behaves more like a particle or as an electromagnetic wave. Explain/justify your choice...

Problem 4 F-18 Production for PET Applications (5 points)

F-18 is a β^+ emitter and it is one of the primary isotopes used for generating positron emission tomography (PET) images for medical diagnostics. F-18 is usually produced in a cyclotron by bombarding O-18 with high energy protons. For production of F-18 via the $^{18}_8\text{O}(p, n)^{18}_9\text{F}$ reaction, determine the minimum allowable energy of the incident protons.

Problem 5 Neutron Reactions in Al27 (10 points)

Neutron bombardment of Al27 can lead to several different neutron reactions. In particular, for the following specific reactions of the form $a(b,c)d$, determine the reaction product d and the Q-value for the reaction:

- a. $^{27}_{13}\text{Al}(n, \gamma)?$ ($\gamma \rightarrow$ gamma ray)
- b. $^{27}_{13}\text{Al}(n, p)?$ ($p \rightarrow$ proton)
- c. $^{27}_{13}\text{Al}(n, d)?$ ($d \rightarrow$ deuteron = deuterium nucleus)
- d. $^{27}_{13}\text{Al}(n, t)?$ ($t \rightarrow$ triton = tritium nucleus)
- e. $^{27}_{13}\text{Al}(n, \alpha)?$ ($\alpha \rightarrow$ alpha particle = He-4 nucleus)

Problem 6 Binding and Separation Energies for O-16 vs. O-17 (10 points)

- a. Determine the binding energy per nucleon for O-16 and O-17.
- b. Determine the neutron separation energy for O-16 and O-17.

From the above computations, what can you say about the stability of O-16 vs. O-17?